A Diastereospecific Synthesis of 2-Methyl-5β-phenyl-5α-carbethoxy-2azabicyclo[2.2.1]heptane: A Ring-Constrained **Analogue of Meperidine**

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Meperidine (1) is an atypical μ -opioid agonist that displays psychostimulant effects¹ that suggest it might also share some pharmacological characteristics with cocaine. Indeed it has recently been shown that, under certain conditions, meperidine will fully substitute for cocaine in squirrel monkeys, in a drug discrimination model of drug abuse.² One hypothesis that has been proposed is that this effect may be due to meperidine interacting at a high affinity binding domain on the dopamine transporter (DAT).^{2,3} The psychostimulant effects of cocaine are also believed to result from its blockade of dopamine uptake. However, in contrast to meperidine, cocaine appears to bind to both high and low affinity sites.⁴ Although the meperidine structure has been studied in relation to its μ -opioid activity,⁵ no work has been carried out to identify the conformation adopted on binding to the DAT.



Meperidine is a 4-phenylpiperidine that has significant structural flexibility and can exist in both chair and boat conformations. To gain a better understanding of the conformation adopted on binding to the DAT, we were interested in studying ring constrained analogues in which the important pharmacophores are held rigidly with respect to one another. Analogues of the chair conformation of meperidine have been prepared and include the tropanes (2a, 2b).⁶ In contrast, no compounds are commercially available that mimic the boat conformation of meperidine. Analogues 3a and 3b have been prepared previously, to study binding to the μ -opioid receptor and appeared ideal for our own studies.⁵ However, we were unable to repeat

5b: R^1 =CO₂Et, R^2 =Ph, R^3 =H



the original synthesis, satisfactorally. Specifically, the detosylation of the 4a, 4b mixture to 5a, 5b could not be performed using the originally reported procedure nor a number of alternative reagents and conditions. This original synthesis also required the separation of the two diastereomers by column chromatography and recrystallization, whereas we now report the diastereospecific synthesis of one of the desired diastereomers using a procedure that is readily reproducible.

trans-L-Hydroxyproline methyl ester (6) is readily made from commercially available *trans*-L-hydroxyproline⁷ which could be converted into 3a in five steps (Scheme 1). Formation of the ethyl carbamate (7) was achieved by treating 6 with ethyl chloroformate and triethylamine in CHCl₃. This was followed by reduction with lithium aluminum hydride (2 equiv) to give the NMe, diol 8. Tosylation using 2.4 equivalents of TsCl in pyridine (4 °C) gave 9 in 18% yield (over the three steps). Alkylation with phenylacetonitrile anion was attempted with three different bases. It was found that best results were obtained with LDA or NaNH₂ in THF, a multicomponent mixture being formed with NaH in DMSO. Thus treatment of 9 with phenylacetonitrile (1.2 equiv) and LDA (2.7 equiv) in THF yielded 10 (48%). Utilization of NaNH₂ in THF led to a 37% yield of 10. Interestingly, these successful alkylations resulted in only one diastereoisomer being formed.⁸ This is in contrast to alkylation utilizing the bulkier *N*-tosyl analogue of **9**, where both isomers are formed (4a, 4b).⁷ This is an unusual finding, in that the smaller group resulted in greater stereoselectivity and may be explained by consideration of the likely transition states 11a and 11b (Figure 1). Clearly the phenyl group has a much greater steric bulk than the nitrile, and thus transition state 11a should be favored over 11b, leading to formation of the exo-phenyl product. However, in the case of the NTs intermediate, edge to face interactions between the two phenyl groups can be envisioned, thus stabilizing the more hindered transition state.

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⁽⁸⁾ The configuration at C-5 was determined by NOE experiments. Irradiation of the H-6 exo, but not the H-6 endo, proton caused an enhancement of the signal for the phenyl ring.

⁽⁹⁾ See the Supporting Information. Newman, A. H.; Kline, R. H.; Allen, A. C.; Izenwasser, S.; George, C.; Katz, J. L. J. Med. Chem. 1995, 38, 3933.

11a:
$$R^1 = Ph, R^2 = CN$$

11b: $R^2 = Ph, R^1 = CN$

Figure 1. Transition states of intermediate nitrile.

This leads to the observed 6:4 ratio of *exo:endo*-phenyl isomers.⁵ In the present work, where the smaller NMe group is utilized, no phenyl–phenyl interactions are possible, leading to specific formation of the *exo*-phenyl product via

the least hindered transition state. Hydrolysis of **10** in 8 N HCl, followed by esterification, gave the desired conformationally restricted analogue of meperidine (3a).⁸

Thus, we have demonstrated a diastereospecific synthesis of the bridged meperidine analogue **3a**. The synthesis of the opposite diastereomer using tosyl-like, phenyl-containing groups and the pharmacological characterization of these compounds are in progress.

Supporting Information Available: Experimental procedures⁹ and spectral data are available for compounds 3a and 6-10 (4 pages).

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